| CYANIDES AMENABLE TO CHLORINATION AFTER DISTILLATION SM 18 th & 20 TH Ed 4500-CN-G | | | | | | | |
|---|-------------------------------|-------------------|---|-----------------|----------|--|--|
| Facility Name: | VELAP ID | | | | | | |
| Assessor Name:Analyst Name: | Analyst Name: | | | Inspection Date | | | |
| Relevant Aspect of Standards | Method Reference | Υ | N | N/A | Comments | | |
| Records Examined: SOP Number/ Revision/ Date Analyst: | | | | | nalyst: | | |
| Sample ID: Date of Sample Preparation: | | Date of Analysis: | | | | | |
| Was calcium hypochlorite solution stored in amber- colored glass bottles in the dark? | 4500-CN ⁻ G 3 c | | | | | | |
| Were two aliquots of each sample taken with one being chlorinated prior to analysis? | 4500-CN ⁻ G 4 a | | | | | | |
| Was calcium hypochlorite solution added to one sample while it was agitated and pH maintained between 11 and 12? | 4500-CN ⁻ G 4 b | | | | | | |
| Was the presence of excess chlorine confirmed after for 1 hour chlorination of the chlorinated aliquot? | 4500-CN ⁻ G 4 b | | | | | | |
| After 1 hour, was the excess chlorine confitmed to be removed from chlorinated aliquot after dropwise addition of NaAsO ₂ or H ₂ O ₂ followed by Na ₂ S ₂ O ₃ ? | 4500-CN ⁻ G 4 c | | | | | | |
| Were samples distilled according to another method and then tested according to another determinative method? | 4500-CN ⁻ G 4 d | | | | | | |
| Was amenable cyanide calculated by subtracting the cyanide found in chlorinated aliquot from the cyanide concentration found in the unchlorinated aliquot? | 4500-CN ⁻ G 5 | | | | | | |
| Notes/Comments: | | | | | | | |